

Particle Physics

Question paper 3

Level	International A Level
Subject	Physics
Exam Board	CIE
Topic	Particle & Nuclear Physics
Sub Topic	Particle Physics
Paper Type	Theory
Booklet	Question paper 3

Time Allowed: 75 minutes

Score: /62

Percentage: /100

A*	A	B	C	D	E	U
>85%	77.5%	70%	62.5%	57.5%	45%	<45%

1 Uranium-236 ($^{236}_{92}\text{U}$) and Uranium-237 ($^{237}_{92}\text{U}$) are both radioactive. Uranium-236 is an α -emitter and Uranium-237 is a β -emitter.

(a) Distinguish between an α -particle and a β -particle.

.....

.....

.....

.....

.....

.....

..... [4]

(b) The grid of Fig. 7.1 shows some proton numbers Z on the x-axis and the number N of neutrons in the nucleus on the y-axis.

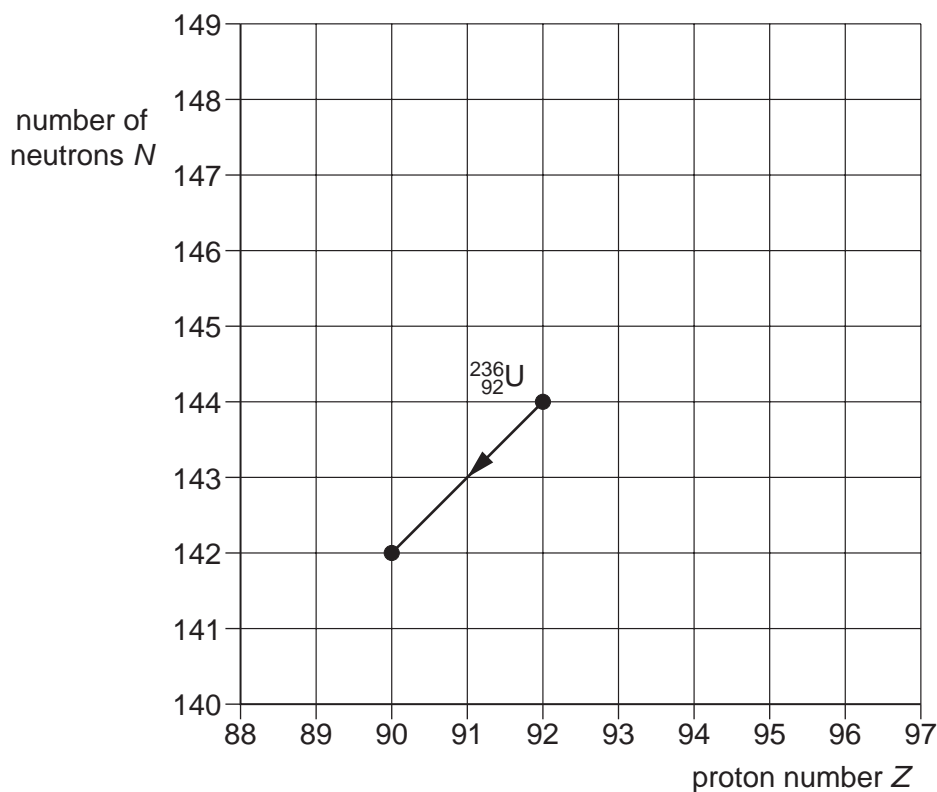


Fig. 7.1

The α -decay of Uranium-236 (${}^{236}_{92}\text{U}$) is represented on the grid. This decay produces a nucleus of thorium (Th).

(i) Write down the nuclear equation for this α -decay.

..... [2]

(ii) On Fig. 7.1, mark the position for a nucleus of

1. Uranium-237 (mark this position with the letter U),
2. Neptunium, the nucleus produced by the β -decay of Uranium-237 (mark this position with the letters Np). [2]

- 2 (a) Evidence for the nuclear atom was provided by the α -particle scattering experiment.
State the results of this experiment.

.....
.....
.....
..... [2]

- (b) Give estimates for the diameter of

(i) an atom,

.....[1]

(ii) a nucleus.

.....[1]

3 The radioactive decay of a strontium (Sr) nucleus is represented in Fig. 7.1.

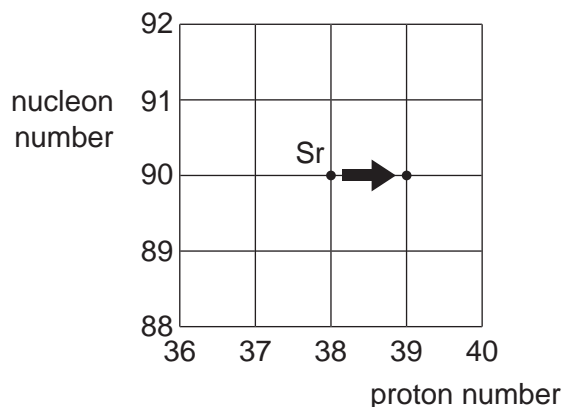


Fig. 7.1

(a) State whether Fig. 7.1 represents α -decay, β -decay or γ -decay.

.....[1]

(b) One type of radioactive decay cannot be represented on Fig. 7.1. Identify this decay and explain why it cannot be represented.

.....
.....
.....[2]

4 The α - particle scattering experiment provided evidence for the existence of a nuclear atom.

(a) State what could be deduced from the fact that

(i) most α -particles were deviated through angles of less than 10° ,

.....
.....
..... [2]

(ii) a very small proportion of the α -particles was deviated through angles greater than 90° .

.....
.....
..... [2]

(b) Fig. 7.1 shows the path AB of an α -particle as it approaches and passes by a stationary gold nucleus.

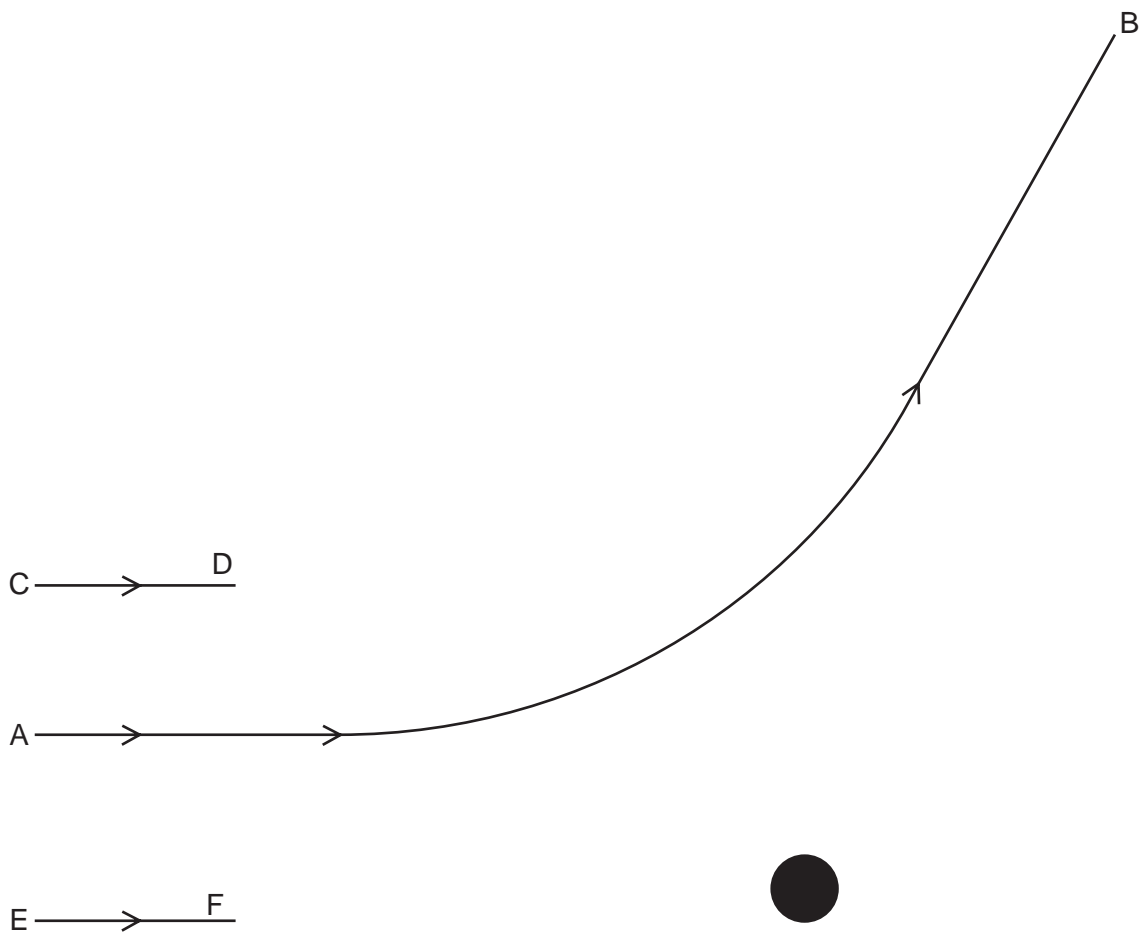


Fig. 7.1

On Fig. 7.1, draw lines (one in each case) to complete the paths of the α -particles passing by the gold nucleus when the initial direction of approach is

- (i) along line CD,
- (ii) along line EF.

[3]

5 One isotope of iron may be represented by the symbol



(a) State, for one nucleus of this isotope,

(i) the number of protons,

number =

(ii) the number of neutrons.

number =

[2]

(b) The nucleus of this isotope of iron may be assumed to be a sphere of radius $5.7 \times 10^{-15} \text{m}$.

Calculate, for one such nucleus,

(i) the mass,

mass = kg

(ii) the density.

density = kg m^{-3}
[4]

- (c) An iron ball is found to have a density of 7900 kg m^{-3} . By reference to your answer in (b)(ii), suggest what can be inferred about the structure of an atom of iron.

.....

.....

..... [2]

6 A nucleus of an atom of francium (Fr) contains 87 protons and 133 neutrons.

(a) Write down the notation for this nuclide.

.....
Fr
..... [2]

(b) The nucleus decays by the emission of an α -particle to become a nucleus of astatine (At).

Write down a nuclear equation to represent this decay. [2]

7 (a) Explain what is meant by an *electric field*.

.....
[1]

(b) A uniform electric field is produced between two vertical metal plates AB and CD, as shown in Fig. 7.1.

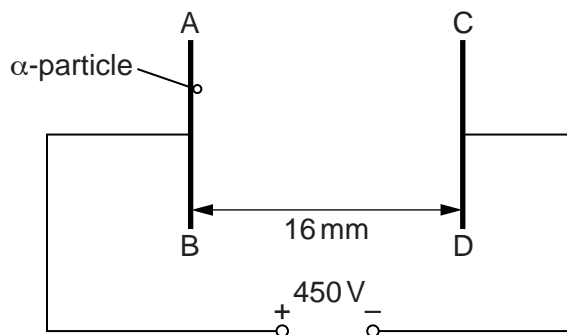


Fig. 7.1

The potential difference between the plates is 450V and the separation of the plates is 16 mm.

An α -particle is accelerated from plate AB to plate CD.

- (i) On Fig. 7.1, draw lines to represent the electric field between the plates. [2]
- (ii) Calculate the electric field strength between the plates.

electric field strength = V m^{-1} [2]

(iii) Calculate the work done by the electric field on the α -particle as it moves from AB to CD.

work done = J [3]

(iv) A β -particle moves from AB to CD. Calculate the ratio

$$\frac{\text{work done by the electric field on the } \alpha\text{-particle}}{\text{work done by the electric field on the } \beta\text{-particle}}$$

Show your working.

ratio = [1]

- 8 (a) An electric field is set up between two parallel metal plates in a vacuum. The deflection of α -particles as they pass between the plates is shown in Fig. 7.1.

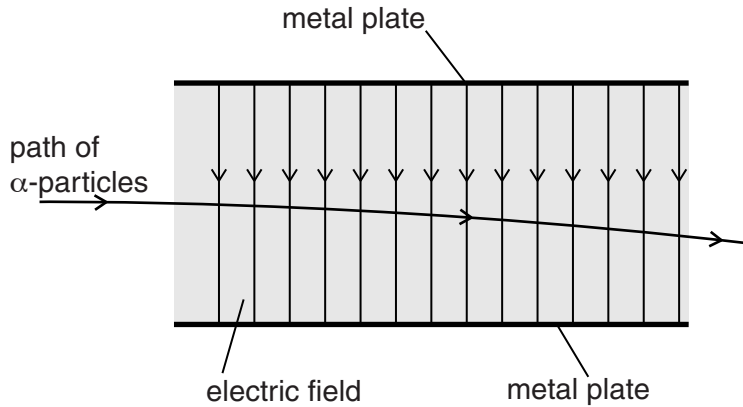


Fig. 7.1

The electric field strength between the plates is reduced. The α -particles are replaced by β -particles. The deflection of β -particles is shown in Fig. 7.2.

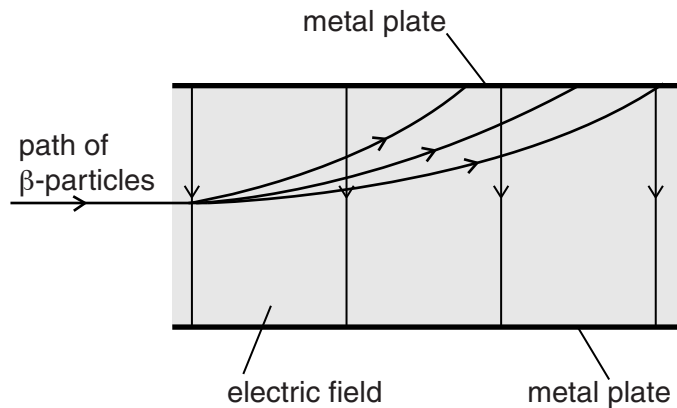


Fig. 7.2

- (i) State one similarity of the electric fields shown in Fig. 7.1 and Fig. 7.2.

.....
 [1]

- (ii) The electric field strength in Fig. 7.2 is less than that in Fig. 7.1. State two methods of reducing this electric field strength.

1.
 2.

[2]

(iii) By reference to the properties of α -particles and β -particles, suggest three reasons for the differences in the deflections shown in Fig. 7.1 and Fig. 7.2.

1.
.....
2.
.....
3.
.....

[3]

(b) A source of α -particles is uranium-238. The nuclear reaction for the emission of α -particles is represented by



State the values of W

X

Y

Z

[2]

(c) A source of β -particles is phosphorus-32. The nuclear reaction for the emission of β -particles is represented by



State the values of A

B

C

D

[1]

- 9 Two horizontal metal plates are separated by distance d in a vacuum. A potential difference V is applied across the plates, as shown in Fig. 6.1.

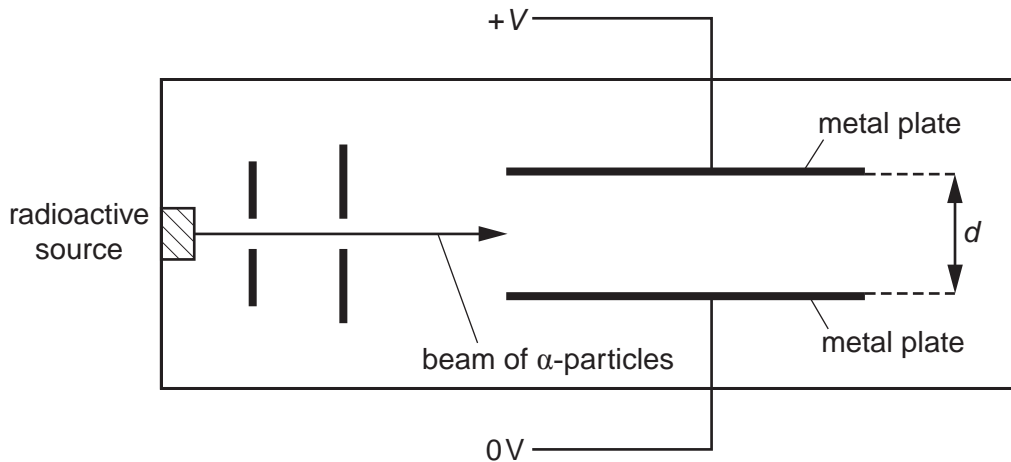


Fig. 6.1

A horizontal beam of α -particles from a radioactive source is made to pass between the plates.

- (a) State and explain the effect on the deflection of the α -particles for each of the following changes:

- (i) The magnitude of V is increased.

.....
..... [1]

- (ii) The separation d of the plates is decreased.

.....
..... [1]

(b) The source of α -particles is replaced with a source of β -particles. Compare, with a reason in each case, the effect of each of the following properties on the deflections of α - and β -particles in a uniform electric field:

(i) charge

.....
.....
..... [2]

(ii) mass

.....
.....
..... [2]

(iii) speed

.....
.....
..... [1]

(c) The electric field gives rise to an acceleration of the α -particles and the β -particles. Determine the ratio

$$\frac{\text{acceleration of the } \alpha\text{-particles}}{\text{acceleration of the } \beta\text{-particles}}$$

ratio = [3]